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| 9 | Methane dio | le laser o | vertone spectr | oscopy at 840 |) nm | | | |
| 11 | | A. Luco | chesini*, S. Gozzin | i | | | | |
| 13 | Istituto per i Processi (| Istituto per i Processi Chimico-Fisici del CNR, Area della Ricerca, Via G. Moruzzi, 1, I-56124 Pisa, Italy Received 22 November 2005, accented 6 February 2006 | | | | | | |
| 15 | 5 | 1.0001/04 25 1.010 | | ., 2000 | | | | |
| 17 | Abstract | | | | | | | |
| 19 | Overtone absorption lines of 12 | Overtone absorption lines of ¹² CH ₄ have been examined by using a tunable diode laser (TDL) spectrometer in the | | | | | | |
| 21 | commercially available heterostruc | n) where the con ture GaAlAs TD | abination overtone band v Ls operating in the "free-ru | $1 + 3v_3$ lies. The spectron inning" mode, which allo | wed the detection | | | |
| 23 | detection technique permitted the | of the line positions within 0.01 cm ⁻¹ . The wavelength modulation spectroscopy (WMS) and the second harmonic detection technique permitted the measurements of minimum absorbances of the order of $\simeq 5 \times 10^{-6}$. This allowed to | | | | | | |
| 25 | $\simeq 0.5 \mathrm{km}^{-1}/\mathrm{amagat}$. For some of temperature. | them self-, air-, | He- and H ₂ -broadening | coefficients have been o | btained at room | | | |
| 27 | © 2006 Published by Elsevier Ltd | | | | | | | |
| 29 | <i>Keywords:</i> Methane absorption coeffic | ients; Overtone ban | ds; Tunable diode-laser spectro | ometer | | | | |
| 31 | 1 Interchection | | | | | | | |
| 33 | | | | | | | | |
| 35 | The ${}^{12}CH_4$ overtones resonances have been detected in the spectra of the outer planets like Urar Neptune, Jupiter, Saturn and relative satellites, in particular Titan, therefore the knowledge of the meth absorption features and their behavior with pressure is of crucial importance for mapping the atmospheres spite of their weakness, overtone and combination tone absorptions are observable in the absorbing layer | | | | | | | |
| 37 | | | | | | | | |
| 39 | the planets because these are much thicker than the ones obtainable in the laboratory. For example the r albedo spectrophotometry measurements made by Karkoschka in 1993 [1] at the European South Observatory show clearly the methane overtones of the visible e.m. spectrum in the jovian planets and also Titanus. | | | | | | | |
| 41 | | | | | | | | |
| 43 | by conventional spectroscopy b | by analyzing the | telescope photographic | plates [2] or by using 1 | ong path White | | | |
| 45 | and Cao [5] took advantage of the intracavity laser spectroscopy (ILS) technique by a sophisticate | | | | | | | |
| 47 | -Freedoordere abbaracae | | | | | | | |

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Here we present a spectroscopic work based on the use of the semiconductor diode lasers (DLs), inexpensive radiation sources still suitable for high-resolution spectroscopy even without any external optical and electrical feedback (*free-running*). Basing on one of the more interesting properties of the DLs, that is the modulability through the injection current, we employed the frequency modulation (FM) technique [6] with the purpose of increasing the signal-to-noise ratio (S/N); this is a necessity when dealing with very weak absorption resonances like the overtones or the combination tones.

When the value of the frequency is chosen much lower than the resonance line-width, the FM spectroscopy is usually called "wavelength modulation spectroscopy" (WMS). In this work, we apply the WMS and the
second harmonic detection to methane around 840 nm, where the CH₄ combination overtone band v₁ + 3v₃ is located.

11

2. Experimental setup

13

Fig. 1 shows the experimental setup adopted for the WMS. The source is a Fabry–Perot-type semiconductor 15 laser Roithner Lasertechnik Mod. RLT85100G that emits up to 100 mW single mode at 847 nm without any external optical feedback, that is in *free-running* mode. It is driven by a stabilized low-noise current supply. Its 17 temperature is monitored by a high-stability controller ($\Delta T = 0.01$ K within 1 h). The temperature control of the DL is very important as its emission wavelength is a linear function of the temperature ($\approx 0.1 \text{ nm/K}$) with 19 periodical mode hops. The wavelength scan is obtained by adding a low-frequency ($\sim 1 \text{ Hz}$) sawtooth signal to the driving current. The wavelength dependence on the current is linear for small current variations $(\approx 0.01 \text{ nm/mA})$, the influence of the quadratic component is less than 10^{-4} nm/mA and is a critical 21 parameter for free-running DL spectroscopy (usual current amplitudes range 10–100 mA). The collimated DL 23 radiation is split in 4 different beams; the first passes through a confocal 5 cm Fabry-Perot interferometer to check the frequency sweep and the laser emission mode; two beams go through two Herriott-type multipass 25 cells 30 m path length each, one containing the sample gas and the other containing water vapor for checking whether the obtained absorption features come from H_2O as expected (see HITRAN database [7]); finally one 27 laser beam passes through an iodine reference cell for the precise wavenumber measurements. The transmitted radiations are collected by pre-amplified silicon photodiodes and the resulting signals are simultaneously



51 Fig. 1. Outline of the experimental apparatus. BS: beam splitter; F.-P.: Fabry-Perot interferometer; M: monochromator; OI: optical insulator; P: photodiode; PC: desk-top computer.

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- 1 acquired by a desk-top computer via a National Instruments Lab-PC 12-bit analog-to-digital converter (ADC). A 35 cm focal length monochromator is employed for the rough wavelength reading ($\Delta \lambda \simeq 0.01$ nm).
- 3 A sinusoidal current is mixed to the DL injection current for the harmonic detection: the signals transmitted through the cells are sent to the lock-in amplifiers to extract the desired harmonic components. An optical
- 5 insulator is put in front of the DL in order to avoid the "feedback noise" due to the reflections coming from the various optical elements along the beam path. The methane gas was supplied by Praxair Inc. with the
- 7 following nominal characteristics: grade 5.5 (purity $\ge 99.9995\%$), H₂O ≤ 2 ppmv,O₂ ≤ 0.5 ppmv, N₂ ≤ 10 ppmv, H₂O ≤ 0.1 ppmv. For accurate pressure measurements in our ranges (from 20 to 0.200 term) we used a connection grave Marian Mad. (542.25.045)

9 300 torr) we used a capacitive gauge Varian Mod. 6543-25-045.

11 **3. Frequency modulation**

We describe the intensity of the radiation after passing through a sample as the product of the incoming intensity $I_0(v)$ and the transmittance $\tau(v)$, where v is the distance from the central frequency of the line:

$$I(\mathbf{v}) = I_0(\mathbf{v})\tau(\mathbf{v}). \tag{1}$$

17 The transmittance in its turn is described by the Lambert–Beer equation:

19
$$\tau(v) = e^{-\sigma(v) z}$$

- 21 where $z = \rho l$ is the column amount (in molecule/cm²), i.e. the product of the absorbing species number density ρ (in molecule/cm³) and the optical path *l* (in cm) of the radiation through the sample; the absorption
- 23 cross-section $\sigma(v)$ is expressed in cm²/molecule. In case of small optical depths [$\sigma(v)z \ll 1$], which is always verified in our case, Eq. (2) can be approximated by

$$\tau(v) \simeq 1 - \sigma(v) z. \tag{3}$$

- 27 $\sigma(v)$ must take into account the shape of the absorption line: Gaussian-like for Doppler broadening and Lorentzian-like for collisional broadening.
- In our measurement conditions other effects like the Dicke narrowing that occurs when the molecular mean free path is comparable to the wavelength of the radiation [8], are not significantly observed and are not taken
- 31 into account. In this scenario, to describe well the absorption coefficient as a function of the radiation frequency is the Voigt function [9], that is the convolution of the Lorentz and the Gauss curves:

35

25

$$f(v) = \int_{-\infty}^{+\infty} \frac{\exp[-(t - v_o)^2 / \Gamma_G^2 \ln 2]}{(t - v)^2 + \Gamma_L^2} \, \mathrm{d}t,\tag{4}$$

- where v_o is the gas resonance frequency, Γ_G and Γ_L are the Gaussian and the Lorentzian half-widths at halfmaximum (HWHM), respectively.
- ³⁹ When FM is used, the emission frequency of the DL is sinusoidally modulated at frequency $v_m = \omega_m/2\pi$ via the injection current,

41
$$v = \bar{v} + a \cos \omega_{\rm m} t. \tag{5}$$

43 If the DL emission frequency \bar{v} is swept over an interval across the chosen transition, one gets a signal that depends on both the line shape and the modulation parameters. Since it is an even function of the time, it can be written as a cosine Fourier series,

47
$$\tau(\bar{v} + a \cos \omega_{\rm m} t) = \sum_{n=0}^{\infty} H_n(\bar{v}, a) \cos n\omega_{\rm m} t, \tag{6}$$

- 49 where $H_n(\bar{v})$ is the *n*th harmonic component of the modulated signal. By demodulating the signal with a lockin amplifier at a multiple nv_m (n = 1, 2, ...) of the modulation frequency, an output signal that is proportional
- 51 to the *n*th component $H_n(\bar{v})$ is collected. When the amplitude *a* is chosen smaller than the width of the transition line, the *n*th Fourier component is proportional to the *n*-order derivative of the original signal,

(1)

(2)

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$$H_{n}(\bar{v},a) = \frac{2^{1-n}}{n!} \left. a^{n} \frac{\mathrm{d}^{n} \tau(v)}{\mathrm{d} v^{n}} \right|_{v=\bar{v}}, \quad n \ge 1.$$
(7)

During this work we used a low modulation amplitude and we detected the second harmonic component (2f detection), which enhanced the S/N ratio and reduced to zero the unwanted background. In order to extract the line parameters, we used a nonlinear least-squares fit procedure explained elsewhere [10]. In particular, to obtain the collisional (pressure) broadening coefficients we fitted the full-width at half-maximum (FWHM) $\gamma_{\rm L}$ vs. pressure by the general expression:

$$\gamma_{\rm L}(p) = 2\Gamma_{\rm L}(p) = \gamma_i p_i + \gamma_{\rm self} p_o,$$

(8)

where γ_i is the FWHM broadening coefficient related to the *i* buffer gas, p_i is the buffer gas partial pressure, γ_{self} is the sample gas FWHM self-broadening coefficient and p_o is the sample gas partial pressure.

13

15

9

4. Experimental results

During this work 48 absorption lines have been observed and their position measured within 0.01 cm^{-1} by the comparison with a very precisely known I₂ absorption spectrum [11] coming from an I₂ reference cell. Table 1 shows the results with the wavenumbers (v') in vacuum. The listed wavelengths are in air at 294 K and

¹⁹ have been deduced by using the index of refraction formula from the work of Edlén [12]. They presumably belong to the combination overtone band $v_1 + 3v_3$ [3], where v_1 is the symmetric stretch and v_3 is the ²¹ asymmetric stretch fundamental vibration.

Ought to the derivative spectroscopy method adopted here it is difficult to obtain absolute intensity measurements directly from the 2*f* signals. We tried to measure the absolute transmission values by using the direct absorption (DA) technique at room temperature (RT) on the same path-length as the one adopted with WMS and with a CH₄ pressure around 95–100 torr, which gives the best S/N. The maximum absorption cross-

whis and with a Chi pressure around 75-100 ton, which gives the best 5/10. The maximum absorption cross-

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|---|---|
| L | / |

| | Table 1 |
|----|--|
| 29 | List of the observed CH ₄ lines along with the maximum absorption cross-section |

| λ at 294 K (nm) | v' (cm ⁻¹) | $\sigma_{\rm max}~(10^{-24}{\rm cm}^2/{\rm molecule})$ | λ at 294 K (nm) | $v' (cm^{-1})$ | $\sigma_{\rm max}~(10^{-24}{\rm cm}^2/{\rm molecule})$ |
|-------------------------|------------------------|--|-------------------------|----------------|--|
| 838.180 | 11927.40 | 2.84 | 841.894 | 11874.78 | 0.75 |
| 838.208 | 11927.00 | 0.74 | 843.364 | 11854.09 | 0.60 |
| 838.214 | 11926.92 | 0.65 | 843.384 | 11853.80 | 1.04 |
| 838.219 | 11926.84 | 1.16 | 843.400 | 11853.58 | 0.25 |
| 838.269 | 11 926.13 | | 843.452 | 11852.85 | 1.30 |
| 838.477 | 11923.17 | 1.14 | 843.495 | 11852.25 | 0.53 |
| 838.755 | 11919.23 | 0.76 | 843.852 | 11847.23 | 0.94 |
| 839.058 | 11914.92 | 3.27 | 843.869 | 11846.99 | 0.61 |
| 839.066 | 11914.81 | | 8443.57 | 11840.15 | 1.71 |
| 839.076 | 11914.66 | 1.57 | 844.748 | 11834.67 | 0.37 |
| 839.955 | 11902.20 | 0.78 | 844.757 | 11834.54 | 1.01 |
| 839.963 | 11902.08 | 1.14 | 845.063 | 11830.25 | 1.26 |
| 839.972 | 11901.95 | 1.19 | 845.077 | 11830.06 | 0.56 |
| 840.087 | 11 900.33 | 0.80 | 845.080 | 11830.01 | |
| 840.123 | 11 899.82 | | 845.629 | 11822.33 | 0.52 |
| 840.242 | 11898.13 | 1.19 | 845.654 | 11821.99 | 0.53 |
| 840.262 | 11897.85 | 0.74 | 845.668 | 11821.79 | 0.48 |
| 840.289 | 11897.47 | 0.81 | 845.677 | 11821.67 | |
| 841.038 | 11886.87 | 1.40 | 845.685 | 11821.55 | |
| 841.056 | 11886.61 | | 845.701 | 11821.33 | |
| 841.178 | 11884.89 | 2.38 | 845.716 | 11821.12 | |
| 841.658 | 11878.12 | 0.43 | 845.737 | 11 820.83 | 1.55 |
| 841.721 | 11877.22 | 1.71 | 845.748 | 11 820.67 | 0.62 |
| 841.840 | 11875.55 | 0.78 | 846.696 | 11807.44 | 0.21 |

4

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- 1 sections shown in the table for most of the lines are obtained by this way, and the measurement errors could not be better than approximately 10%.
- 3 An example of the spectrum as obtained by WMS and the second derivative technique is shown in Fig. 2, where four methane lines are extracted by the fit procedure. When dealing with free-running DLs, a frequency
- 5 sweep of their emission is always associated with a variation of the emission intensity, therefore in Fig. 2 the relative intensity ratios does not give the real ratios, but modified by this intensity change. In particular in this
- 7 case, when moving the LD emission wavenumber by -1 cm^{-1} , the LD emission intensity increased by 3.8 times.
- 9 The complexity of the structure of the $v_1 + 3v_3$ band does not permit an immediate and correct quantum classification of the ro-vibrational resonances. For these highly excited levels the numerous possible
- 11 resonances between ro-vibrational levels can modify significantly the intensity as well as the position of the expected lines. Tentatively, basing upon the intensity distribution, the lines within $(11\,807.44 \le v' \le$
- 13 11 854.09) cm⁻¹ should belong to the *P* branch, the ones within $(11874.78 \le v' \le 11886.87)$ cm⁻¹ to the *Q* branch and the lines in between $(11897.47 \le v' \le 11927.40)$ cm⁻¹ to the *R* branch.
- 15

4.1. Line broadening and shifting measurements

- Table 2 shows the measured CH₄ pressure broadening coefficients by itself, air, H₂ and He gases, all at RT for some of the more intense observed lines. The shown errors are the maximum errors (3σ) .
- In order to obtain the pressure shift coefficients we acquired simultaneously the CH_4 and I_2 absorptions from their respective measurements cells, taking the latter at fixed pressure and temperature for reference; sometimes we used H_2O at fixed pressure at RT in its cell as the reference.
- Examples of pressure broadening and shifting measurements are in Figs. 3 and 4, where CH₄ broadening and shifting behaviors by collision with itself and helium buffer gas are displayed, respectively, with their best linear fit.
- For the 840 nm band there are no previous pressure broadening and shifting measurements therefore a comparison can be done only with the results on different wavelengths. Moreover ought to the lack of quantum assignment of the lines in this band, only an average comparison can be done. Most of our
- ²⁹ quantum assignment of the lines in this band, only an average comparison can be done. Most of our broadening coefficients at RT in average are a little lower than the ones coming from previous works on $2v_3$ overtone [13,14]. The same happens for the self- and the H₂-broadening [15], for He- and H₂-broadening for



Fig. 2. Second derivative signal of the methane transmission spectrum around 838.2 nm (a) obtained by WMS with 10 Hz bandwidth at $p_{CH_4} = 95$ torr and T = 296 K. The peaks extracted by the fit procedure (b) are shifted down for clearness.

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| 1 | Table 2 |
|---|---|
| | Methane pressure broadening FWHM coefficients (v) at room temperature |

| $v' (cm^{-1})$ | γ_{self} (MHz/torr) | γ_{air} (MHz/torr) | γ_{H2} (MHz/torr) | $\gamma_{\rm He}~({\rm MHz/torr})$ |
|--------------------|--|---------------------------|--------------------------|------------------------------------|
| 11914.92 | 6.4 ± 0.2 | 4.5 ± 0.2 | 4.8 ± 0.1 | 3.6 ± 0.1 |
| 11898.13 | 6.1 ± 0.1 | 3.4 ± 0.2 | 4.8 ± 0.2 | 3.4 ± 0.2 |
| 11 897.85 | 5.4 ± 0.7 | 3.4 ± 0.4 | 3.6 ± 0.5 | 3.0 ± 0.8 |
| 11886.87 | 6.4 ± 0.1 | 4.4 ± 0.4 | 5.0 ± 0.2 | 3.7 ± 0.3 |
| 11853.80 | 6.5 ± 0.1 | 6.2 ± 0.3 | 4.8 ± 0.2 | 3.6 ± 0.3 |
| 11847.23 | 8.7 ± 0.2 | 6.3 ± 0.6 | 8.5 ± 0.2 | 3.8 ± 0.9 |
| 11840.15 | 5.2 ± 0.1 | 3.7 ± 0.3 | 4.9 ± 0.2 | 3.2 ± 0.1 |
| Lorentz FWHM (GHz) | , e, | ´3 | 8.2 | |
| 0 + + + + + + | | | -4 | |
| 0 (a) | 50 100 150 Pressure (Torr) | 200 250 (b) | 0 50 100 Pressure | 150 200 250 e (Torr) |

Fig. 3. Variation of the Lorentz width (a) and of the relative line position Δv (b) in GHz of the 11 886.87 cm⁻¹ absorption line as a function of CH₄ pressure at room temperature.



45

Fig. 4. He-broadening (a) and -shifting (b) measurements for the 11840.15 cm⁻¹ methane absorption line (room temperature).

the 3v₃ overtone band [16] and for the air-broadened v₃-fundamental [17], while there are no much differences
in the H₂- and He-broadening measurements on the v₄-fundamental [18] at RT. As already said, our system has not enough sensitivity to estimate the contribution of an eventual Dicke narrowing, but this phenomenon could explain some of these differences.

The summary of the measured pressure shift coefficients is shown in Table 3. The air-shift coefficients are in average roughly 10 times higher than the ones got by others on the fundamentals v_2 and v_4 [19] and very much dependent on the lines.

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| 1 | Table 3 | |
|---|--|---------------------|
| | Methane pressure shift coefficients (δ) | at room temperature |

| 3 | $v' (cm^{-1})$ | $\delta_{ m self}$ (MHz/torr) | $\delta_{\rm air}~({ m MHz/torr})$ | $\delta_{\rm H_2}$ (MHz/torr) | δ_{He} (MHz/torr) |
|---|----------------|-------------------------------|------------------------------------|-------------------------------|-----------------------------------|
| 5 | 11 914.92 | -1.3 ± 0.3 | -1.1 ± 0.1 | -0.7 ± 0.3 | 0.3 ± 0.1 |
| | 11 898.13 | -1.1 ± 0.1 | -0.6 ± 0.1 | -0.6 ± 0.1 | -0.2 ± 0.1 |
| 7 | 11 897.85 | -1.4 ± 0.2 | -0.9 ± 0.1 | -0.7 ± 0.1 | 0.4 ± 0.3 |
| / | 11 886.87 | -1.0 ± 0.1 | -0.5 ± 0.1 | -0.8 ± 0.1 | 0.1 ± 0.3 |
| ~ | 11 853.80 | -0.9 ± 0.1 | -0.7 ± 0.1 | -0.9 ± 0.1 | 0.0 ± 0.1 |
| 9 | 11 847.23 | -0.3 ± 0.4 | -0.8 ± 0.3 | -0.4 ± 0.3 | 0.1 ± 0.3 |
| | 11840.15 | -1.5 ± 0.1 | -0.9 ± 0.1 | -0.8 ± 0.1 | 0.0 ± 0.2 |
| 1 | | | | | |

11

13 Very recently Predoi-Cross [20] et al. measured the self-broadening and shift coefficients in laboratory by using the McMath–Pierce Fourier transform spectrometer of the Kitt Peak National Solar Observatory in the

15 $v_1 + v_4$, $v_3 + v_4$ and $v_2 + v_3$ combination tone bands of ¹²CH₄: while the self-broadenings are quite the same, the self-shifts are two or three times lower than ours. The same happens in the comparison with the recent 17 work on the $v_2 + v_4$ combination band made by Mondelain et al. [21]. These big differences are noted also in

the air- and He-shifts on the R3 triplet in the $2v_3$ band by Zèninari et al. [14], where in particular they show a linear dependence of the air-induced shift with the energy of the upper vibrational state: this occurs because

the shift value depends on the vibrational quantum numbers as well as the polarizability of the perturbing 21 element [22].

23 5. Conclusion

By diode laser WM spectroscopy with high resolving power ($\lambda/\Delta\lambda \approx 10^7$) and the aid of a 30 m total pathlength multipass Herriott-type measurement cell, 48 CH₄ absorption lines have been detected around 11 900 cm⁻¹ and their positions measured within 0.01 cm⁻¹. They presumably belong to the combination

overtone $v_1 + 3v_3$ ro-vibrational band. The absolute line positions have been obtained by comparison with a reference I₂ absorption cell and the utilization of a very precise atlas. The maximum absorption cross section

of the observed lines ranges between 2×10^{-25} and 3×10^{-24} cm²/molecule at RT. Collisional broadening and shift coefficients for different perturbing gases have been measured at RT for 7 of the more intense absorption lines.

33

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